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(1) Chapter 6, Problem 11

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When 2.06 grams of titanium is combusted in a bomb calorimeter, the temperature of the calorimeter increases from 25.0°C to 91.6°C. Calculate the heat of reaction for the combustion of one mole of titanium metal. The heat capacity of

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the calorimeter is $9.84 \text{ kJ/}^\circ\text{C}$. A) 14.3 kJ B) 19.6 kJ C) -318 kJ D) $-1.52 \times 10^4 \text{ kJ}$

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Multiple Choice 10 Terms. hanne1126; Subjects. Arts and Humanities. Languages. Math.

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Question.7. Calculate the Physical Fitness Index for a 12 year old boy having completed Harvard step test for a duration of 3 min and a pulse rate of 54 beats for 1 to 1.5 min, 50 beats for 2 to 2.5 min and 46 beats for 3 to 3.5 min. Answer. 3 Marks

Questions Question.8. List the six items of 'AAHPER' motor fitness test. Answer.

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