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REVIEW SHORT
ANSWER Answer the

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following questions in the space provided. 1. b A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

8 Chemical Equations and Reactions

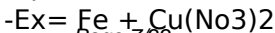
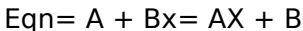
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-One element in the reactant of a chemical compound replaces one element in the product of a Chemical Reaction -General



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Modern Chemistry 1

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Notes and Answers

Chapter 8 SECTION 1

SHORT ANSWER 1. a. d

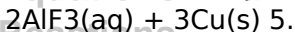
b. a c. b d. f e. e f. c 2.

8,4,9 3. a. 12 atoms b.

16 atoms c. 51 atoms

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d. 3×10^{24} atoms 4.



a.

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chemical equation, you
can obtain the 1.

Chemical formulas of
the reactants and
products. 2. Relative
amounts of the

reactants and products

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3. Physical state of the reactants and products. 4. All of the above.

Answers

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This results in a correct chemical equation, or balanced formula equation, for the burning of methane in oxygen. $\text{CH}_4(\text{g}) + 2\text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{g})$ This reaction is further illustrated in Figure

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8-3. Additional Symbols
Used in Chemical
Equations Table 8-2 on
page 246 summarizes
the symbols commonly
used in chemi-
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AND REACTIONS + 2 +
+ +

CHAPTER 8 Chemical Equations and Reactions

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Chemical Equations

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Chemical
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1 SHORT ANSWER

Answer the following questions in the space provided. 1. Match the symbol on the left with its appropriate description on the right. → (a) A precipitate forms. ↓ (b) A gas forms. ↑ (c) A reversible reaction occurs.

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CHAPTER 8 REVIEW.

Chemical Equations and Reactions. MIXED REVIEW. SHORT

ANSWER Answer the following questions in the space provided. (1 point) 1. _____ A balanced chemical equation represents all the following . except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

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SECTION 8.1 REVIEW.
SHORT ANSWER

Answer the following
questions in the space
provided. 1. Match the
symbol on the left with
its appropriate
description on the
right.

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Chemical Equations
and Reactions The
evolution of energy as
light and heat is an
indication that a
chemical reaction is
taking place. CHAPTER
8 Fireworks. CHEMICAL
EQUATIONS AND
REACTIONS 261

SECTION 1 O

OBJECTIVES List three
observations that
suggest that a

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chemical reaction has
taken place.

**CHAPTER 8 Chemical
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Chemical reactions are described by chemical equations. A chemical equation represents, with symbols and formulas, the identities and relative amounts of the reactants and

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products in a chemical reaction.

Reactions
Answers
**[HOT!] Chapter 8
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Katalynn Sutton
Chapter 8 Review
Chemical Equations 8.1
A chemical equation is a shorthand expression for a chemical change or reaction which uses chemical symbols and formulas of the reactant and products

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Reactions

and other symbolic terms to represent a chemical reaction.

Answers **Chapter 8 Review - Katalynn Sutton Chapter 8 Review ...**

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Practice. Day 9
(1/16/20) - Test:
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Chapter Eight

[Chemical Equations
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Nine [Stoichiometry]

Chapter Ten [States of
Matter] Chapter Eleven

[Gases] Chapter

Twelve [Solutions] ...

Section 2: Chapter
review 18 thru 21.

Section 3: Chapter
review 31 thru 33.

Homework Answers .

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Chapter Eight [Chemical Equations and Reactions]

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Class: Date: Chemical
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SHORT ANSWER

Answer the following
questions in the space
provided. 1. Match the
symbol on the left with
its appropriate
description on the
right. e- (aq) (a) A
precipitate forms. (b) A

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gas forms. (c) A reversible reaction occurs. (d) Heat is applied to the reactants.

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