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Learning Outcomes
Upon completion of
this lab, the student
will be able to: 1)
Examine different

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types of chemical reactions. 2) Express chemical equations in molecular, ionic, and net-ionic forms. 3) Predict solubility rules for some ionic substances in water
Introduction There are several different types of chemical reactions such as: combustion ...

Experiment A5- Reactions - 1 EXPERIMENT A5 TYPES OF ...

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$I^{-}(aq) + BrO_3^{-}(aq) + 6 H^{+}(aq) \rightarrow 3 I_2(aq) + Br^{-}(aq) + 3 H_2O(l)$ The Reaction Rate Depends On The Concentration Of Reactants ...

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EXPERIMENT A7

continued 5.

Organizing Ideas Write
the equation for the
reaction that occurred

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when you heated hydrated $MgSO_4$ in this experiment. Use the letter n to represent the number of moles of water driven off per mole of anhydrous magnesium sulfate. 6. Organizing Conclusions Using your answers to Calculations items 2,

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