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like in hydrogen, that has only one electron in the orbital. a full octet wouldn't affect the activity of these metals. (if anyone has a different answer please send it to me) why does hydrogen act like a group 17 element when forming covalent bonds. its not

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stable by itself so it bonds quickly.

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together by. Chapter 6 Test DRAFT. 10th grade. 527 times. Science. 83% average accuracy. a year ago. joecarter. 0. Save. Edit. Edit. ... answer choices . one atom of chlorine. two atoms of chlorine. one atom of oxygen.

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1. Use the full name of the cation (first ion) 2. Add the suffix -ide

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to the name of the anion (second ion) if it is monatomic (Use the full name of the anion if it is polyatomic) \*Include the roman numeral if there is a transition metal. Old System. -ous for lower charged ion, -ic for higher charged ion.

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Chemistry Chapter 6 Review. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. emma\_pear. Terms in this set (32) Identify and define the three major types of chemical bonding. In ionic bonding, large numbers of oppositely charged ions join because of mutual electrical

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Chemistry (4th Edition) Burdge, Julia Publisher McGraw-Hill Publishing Company ISBN 978-0-07802-152-7

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\_\_\_ 6. A repetition of properties occurs when elements are arranged in order of increasing atomic number. \_\_\_ 7. vertical column in the periodic table Multiple Choice Identify the choice that best completes the statement or answers the question. \_\_\_ 8. Which of the following elements is in the same period as phosphorus? a. magnesium c. nitrogen

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