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[2] (b) Marit arranged the values she found in Table 12 into an energy cycle. $C_2H_4(g) + H_2(g)$
 $C_2H_6(g)$ $3 O_2$ $2 CO_2(g) + 3H_2O$ - 1 4 1 1 k J m o l⁻¹ - 2 8 6 k J m o l⁻¹ - 1 5 6 0 k J m o l⁻¹
(hydrogenation) (l) Calculate the value for the enthalpy of hydrogenation of ethene from the energy cycle.....

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Chemistry SL Paper 2 TZ1 Markscheme 2 Topics: Covalent bond , Hydrogen bond , Yield Pages: 13 (2237 words) Published: March 3, 2015

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8 20 2 2 2 2 o correct reactants and products; [2] correct coefficients; M2 can only be scored if M1 correct. (b) Local pollutant: carbon monoxide/CO / volatile organics/VOCs / nitrogen oxide/NO / (unburnt) hydrocarbons; Do not accept methane/CH₄, ethane/C₂H₆, propane/C₃H₈ or butane/C₄H₁₀. Global pollutant: nitrogen oxide/NO / carbon ...

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$CH_3CHOH(l)$ $3O_2(g)$ $2CO_2(g)$ $3H_2O(l)$ $-1367 kJ$ $3 2 + \rightarrow + 2 2 2 H = mol \ddot{O} 1 -1 CH (l) 1 2O(g) 8CO(g)$
 $9HO(l)$ $8 18 + \rightarrow + 2 2 2 2 H = -5470 kJ mol \ddot{O} (i)$ Calculate the mass, in g, of ethanol and octane in 1.00 kg of the fuel mixture. [1].....

CHEMISTRY Candidate session number STandaRd lEveI PaPER 2

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2209-6110 - 3 - Turn over M09/4/CHEMI/SPM/ENG/TZ1/XX+ 1. The molar mass of a compound is

approximately 56 g mol⁻¹. Which formula is possible for this

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possible liver damage; [2 max] (b) mild analgesics function by intercepting the pain stimulus at the source / interfere with the production of substances that cause pain/prostaglandins; strong analgesics work by bonding to receptor sites in the brain / prevent the transmission of pain impulses without depressing the central nervous system; [2]

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