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Answer: $\rho = 737 \text{ kg/m}^3$; 1 gal = 3.788 L = $3.788 \times 10^{-3} \text{ m}^3$. The mass of a ^{235}U nucleus is 235mp, then. (a) for 1 gallon, $m = \rho V = (737 \text{ kg/m}^3)(3.788 \times 10^{-3} \text{ m}^3) = 2.79 \text{ kg} = 2.79 \times 10^{-3} \text{ g}$. So, $(1.3 \times 10^8)/(2.79 \times 10^{-3} \text{ g}) = 4.7 \times 10^5/\text{g}$. (b) 1 g contains $(1.00 \times 10^{-3} \text{ kg})/(235 \text{ mp}) = 2.55 \times 10^{21}$ nuclei.

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