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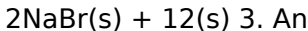
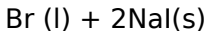
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Note: Not all of the
reactions will occur. For
those that do not, write

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no reaction. 1. A piece of copper is dropped into a container of water. No reaction 2.

Liquid bromine is added to a container of sodium iodide crystals.



3. An aluminum strip is immersed in a solution of silver nitrate.

Ultimate equation answers - Max Study

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Handbook by
Hague and Smith 7.

Free elements, no
matter how complex
the molecule, have an
oxidation number
(valence or charge)
equal to zero. Oj. Fe.

T~ He < /1 1 ~ ~ L-td.

8. The following are
diatomic or polyatomic
elements in nature
which must be
committed to memory.

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answers chapter 10
compilations from
roughly the world.
when more, we here
come up with the
money for you not by
yourself in this kind of
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Pages are from the
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Precipitation
Reactions(See

solubility rules) - Pg.

28, Practice-Pg. 49 •

Formation of a Gas -

Pg. 50, Practice-Pg. 51

• Acid Base

Neutralization

Reactions- Pg. 53,

Practice-Pg.54 oWeak

acids and bases do not

break apart and are in

net ionic equation. •

Synthesis and

Decomposition

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Reactions - Pgs. 42-43,
Practice-Pg. 43 •
Single Replacement
Reactions - Pg. 45,
Practice-Pg. 46.

Quiz on all Chemical Reactions - Max Study

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SnCO tin(II) carbonate
sodium hydrogen
carbonate NaHCO 1. 2.
3. 4. 5. 6. 8. 9. 10. 11.
vanadium(V) oxide H₂O
dihydrogen monoxide

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(NH) CO ammonium
oxalate 4 224
polonium(VI)
thiocyanate Po(SCN)₆
tetrphosphorus
decaoxicle P O 4 10 12.
13. 15. 16. 17. 18. 19.
manganese(VII) oxide
copper(II) dihydrogen
phosphate

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Answers Chapter
3
Correlating questions

(ii) from: Rodney
Schmitz (2007) Moore
County Schools, NC
Reaction Prediction – 5

For each of the
following eight
reactions, in part (i)
write a BALANCED
equation and in part (ii)
answer the question
about the reaction.

KEY to 5 Reaction

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Prediction -

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Determine the
oxidation number of
the underlined
element: KMnO_4 Since
K is an alkali metal, its
charge must be $1+$
Oxygen is $2-$ but there
are four of them,
therefore, 4 times $2-$
equals $8-$ If $1+$ and
 $8-$ are added Ultimate
Chemical Equation

Answers

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Determine the
oxidation number of
the underlined
element: KMnO_4 Since
K is an alkali metal, its
charge must be $1+$
Oxygen is $2-$ but there
are four of them,
therefore, 4 times $2-$

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equals 8— If 1+ and
8— are added The
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